JASMONIC ACID-LIKE SUBSTANCES FROM THE CULTURE FILTRATE OF BOTRYODIPLODIA THEOBROMAE

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Abstract—Four cyclopentanoidal fatty acids were isolated from the fungus *Botryodiplodia theobromae* and identified as jasmonic acid-like substances. Their structures are (+)-4,5-didehydro-7-iso-jasmonic acid [(1R, 2S)-(+)-3-oxo-2-(2Z-pentenyl)cyclopent-4-en-1-yl-acetic acid], ethyl <math>(+)-7-iso-jasmonate [ethyl (1R, 2S)-(+)-3-oxo-2-(2Z-pentenyl)cyclopent-1-yl-acetate], <math>(1R, 2S)-(+)-3-oxo-2-(2Z-pentenyl)cyclopent-1-yl-propionic acid and <math>(1S, 2S)-(+)-3-oxo-2-(2Z-pentenyl)cyclopent-1-yl-butyric acid.

INTRODUCTION

Botryodiplodia theobromae Pat. is a common tropical fungus known to produce several cyclopentanoidal fatty acids with plant growth regulating activities [1,2]. Major substances are (+)-7-iso-jasmonic acid (1) and (-)-jasmonic acid (5). Some other metabolites possessing the 7-iso-configuration have already been described as (+)-11,12-didehydro-7-iso-jasmonic acid, (+)-9,10-dihydro-7-iso-jasmonic acid and cucurbic acid [2].

In continuation of the phytochemical investigation of this fungus we now report the isolation of four further compounds, one of which is new and the three others described for the first time as metabolites of *Botryodiplo*dia theobromae.

RESULTS AND DISCUSSION

From a surface culture of *Botryodiplodia theobromae* three acidic substances and one neutral compound were isolated and after chromatographic purification by CC, TLC and HPLC, identified by GC-MS, IR, optical rotation and ¹H NMR, respectively, combined with chemical modifications. The acidic extract contained in low yields (+)-4,5-didehydro-7-iso-jasmonic acid (9), (1R,2S)-(+)-3-oxo-2-(2Z-pentenyl)cyclopent-1-yl-propionic acid (11) and (1S,2S)-(+)-3-oxo-2-(2Z-pentenyl)cyclopent-1-yl-butyric acid (12); the neutral compound was proved to be ethyl (+)-7-iso-jasmonate (2).

The identity of 2 was confirmed by GC-MS and comparison with authentic (-)-6. The key fragment at m/z 173 ([M-OC₂H₅]⁺) represents the unique difference to the mass spectrum of 11-methyl ester. Besides 2, compound 6 was present as indicated by GC, but its natural occurrence has to be doubted because possible isomerization of 2 to 6 could not be excluded. The unsaturated acid 9 is accompanied by 10 in the ratio 4:1 as shown by TLC. Hydrogenation of 9-10 with sodium borohydride, reducing both the keto groups and the ring double bonds, gave the four isomers 3, 4, 7 and 8 [3].

R1
R1 = O, R2 = H
R1 = O, R2 = H
R2
R1 = O, R2 = C2H
R3
R1 =
$$\alpha$$
-OH, β -H, R2 = H
R1 = α -H, β -OH, R2 = H
R2
R1 = α -H, β -OH, R2 = H
R3
R1 = α -H, β -OH, R2 = H
R4
R1 = α -H, β -OH, R2 = H
R5
R1 = α -H, β -OH, R2 = H
R6
R1 = α -H, β -OH, R2 = H
R7
R1 = α -H, β -OH, R2 = H
R8
R1 = α -H, β -OH, R2 = H
R9
R1
R1 = α -H, β -OH, R2 = H
R1
R1 = α -H, β -OH, R2 = H
R1
R1 = α -H, β -OH, R2 = H
R1
R1 = α -H, β -OH, R2 = H
R2
R1 = α -H, β -OH, R2 = H
R3
R1 = α -H, β -OH, R2 = H
R4
R1 = α -H, β -OH, R2 = H
R5
R1 = α -H, β -OH, R2 = H
R6
R1 = α -H, β -OH, R2 = H
R7
R1 = α -H, β -OH, R2 = H
R8
R1 = α -H, β -OH, R2 = H
R1
R1 = α -H, β -OH, R2 = H
R1
R1 = α -H, β -OH, R2 = H
R1 = α -H, β -OH, R2 = H
R1 = α -H, β -OH, R2 = H
R1 = α -H, β -OH, R2 = H
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R1 = α -H, β -OH, R2 = H
R1 = α -H, β -OH, R2 = H
R1 = α -H, β -OH, R2 = H
R1 = α -H, β -OH, R2 = H
R1 = α -H, β -OH, R2 = H
R1 = α -H, β -OH, R2 = H
R1 = α -H, β -OH, R2 = H
R1 = α -H, β -OH, R2 = H
R1 = α -H, β -OH, R2 = H
R1 = α -H, β -OH, R2 = H
R1 = α -H, β -OH, R2 = H
R1 = α -H, β -OH, R2 = H
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R1 = α -H, β -OH, R2 = H
R1 = α -H, β -OH, R2 = H
R1 = α -H, β -OH, R2 = H
R1 = α -H, β -OH, R2 = H
R1 = α -H, β -OH, R2 = H
R1 = α -H, β -OH, R2 = H
R1 = α -H, β -OH, R2 = H
R1 = α -H, β -OH, R2 = H
R1 = α -H, β -OH, R2 = H
R1 = α -H, β -OH, R2 = H
R1 = α -

Catalytic hydrogenation of the methyl esters gave 9,10-dihydro-1 methyl ester and 9,10-dihydro-5 methyl ester [2,4,5]. Like 6, compound 10 might be an artifact originating from 9 by isomerization during the isolation procedure.

In the mass spectrum of 11-methyl ester the fragments $[M-(CH_2)_2COOMe]^+$ and $[M-C_5H_8]^+$ indicate a *n*-propionic acid side chain at C-1 and a pentenyl side chain at C-2. The base peak at m/z 83 ($[C_5H_7O]^+$) stems from the cyclopentanone ring [6]. The *cis*-configuration of

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both side chains was deduced from the positive optical rotation [7] and verified by the facile alkaline isomerization to 13. The 1H NMR spectrum, proving the cisdouble bond, was very similar to that of 5 [6]. Hydrogenation with Adams catalyst of 13-methyl ester introduced two hydrogens in the pentenyl side chain as shown by the key ion at m/z 170 ([$M-C_5H_{10}$] $^+$) in the mass spectrum. The 12-methyl ester showed a similar fragmentation pattern as described for [^{18}O]-12-methyl ester [8] with the key fragments at [$M-(CH_2)_3COOMe$] $^+$ (n-butyric acid side chain), [$M-C_5H_8$] $^+$ (pentenyl side chain) and the typical base peak at m/z 83. The facile chemical isomerization of 12 to 14 confirmed a cis-configuration of both side chains. Compound 12 is dextrorotatory, therefore, it has (1S, 2S)-configuration [7].

The presence of 9 in the fungus Botryodiplodia theobromae is understandable by its biogenetic formation from 12-oxo-phytodienoic acid via β -oxidation without elimination of the ring double bond [8, 9]. Substance 12 is already known to be a precursor of 7-iso-jasmonic acid (1) [8]. The structure of the new compound 11 suggests, that enzymes of jasmonic acid biosynthesis might be able to convert also a longer unsaturated fatty acid to give this jasmonic acid-like substance, as shown in other examples [10, 11]. Substance 2, probably formed from 1 by fungal enzymes, has already been found in flower oils of Jasminum grandiflorum [12].

EXPERIMENTAL

Chromatographic methods. TLC (silica gel GF_{2.54}): (a) CHCl₃–MeOH–HOAc (140:20:1); (b) n-hexane–EtOAc–(60:40:1); detection by anisaldehyde reagent and heating for 5–10 min at 120° [13]; prep. TLC (silanized silica gel, RP2):(c) C_6H_6 –Me₂CO (17:13). CC: (600 × 20 mm) on silanized silica gel prepared by treatment of silica gel with TMCS in C_6H_6 ; elution with a stepwise gradient of EtOAc in CHCl₃. HPLC (method a): LiChroprep RP 8 (30–63 μ m, 310 × 25 mm), elution with MeOH–0.1% H_3 PO₄ in H_2 O (3:2), flow rate 3 ml/min, UV detector at 228 nm; (method b): Polyol RP 18 (250 × 4.6 mm), elution with MeOH–0.1% H_3 PO₄ in H_2 O (1:1), flow rate 1 ml/min, UV-detector at 228 nm.

Fermentation. The fungus Botryodiplodia theobromae Pat. (strain D7/2, isolated from Cuban oranges, Citrus sinensis Osbeck cv. Valencia) was pre-cultured on malt agar. The mycelium was homogenized in $\rm H_2O$ and used for inoculation of the production medium. The fungus was grown in surface culture for 7 days at 30° in a flask (400 ml) containing 100 ml of the following medium: 30 g sucrose, 5 g soya flour, 15 ml corn steep liquor, 10 ml of mineral salt soln, diluted to 11 with $\rm H_2O$ and to pH 5.4–5.6 with 1 M NaOH and sterilized at 120° for 20 min. The mineral salt soln used contained 0.5 g KH₂PO₄, 0.3 g MgSO₄ · 7 H₂O, 10 mg FeSO₄ · 7 H₂O, and 8.8 mg ZnSO₄ · 7 H₂O/l.

Isolation procedure. The contents from the 100 culture flasks were lyophilized and extracted with EtOAc (3×1.5 l). Acidic compounds were sepd by extraction into satd NaHCO₃ soln (3×150 ml). Neutral compounds (containing 2) remained in the EtOAc layer. Re-extraction of the aq. phase with GHCl₃ (3×150 ml) after acidification to pH 3.5 with 4 M HCl gave a CHCl₃ extract which was dried (Na₂SO₄) and evapd. The residue was divided into three parts and each of it purified by CC. Fractions eluted with CHCl₃-EtOAc (9:1) gave a crude mixture which on prep. HPLC (method a) gave two highly enriched fractions of 9 (R_i : 66-73 min) and 11+12 (R_i : 96-125 min). Fractions were extracted with CHCl₃ after dilution with H₂O and further separated by HPLC (method b) giving:

(+)-4,5-Didehydro-7-iso-jasmonic acid [(1R, 2S)-(+)-3-oxo-2-(2Z-pentenyl)cyclopent-4-en-1-yl-acetic acid] (9). 1.5 mg; R_f in HPLC (method b): 9.2 min; R_f in TLC system a (0.20) and b (0.42); $[\alpha]_D^{25}$ dextrorotatory; MS (80 eV) m/z (rel. int.) of 9-methyl ester: 222 [M]+ (21), 193 [M-C₂H₅]+ (17), 191 [M-OMe]+ (11), 167 (15), 154 [M-C₅H₈]+ (81), 149 [M-(CH₂)COOMe]+ (18), 133 (21), 119 (16), 107 (21), 95 [C₆H₇O]+ (100); ¹H NMR, IR of 9-methyl ester identical with the data published [12].

(1R, 2S)-(+)-Oxo-2-(2Z-pentenyl)cyclopent-1-yl-propionic acid (11). 0.8 mg; R_t in HPLC (method b): 11.7 min; R_f in TLC systems a (0.25) and b (0.50); $[\alpha]_D^{25}$ +55° (MeOH; c 0.8); IR $v_{\max}^{\text{CHCl}_3}$ cm $^{-1}$: 3000, 1734, 1650; 1 H NMR (200.13 MHz, CDCl $_3$, TMS as internal standard): δ 0.93 (3H, t, J_{AB} = 7.3 Hz, -Me), 1.73–2.73 (14H, m), 5.23 (1H, dt, J_{AB} = 10.5 Hz, J_{AX} = 7.0 Hz, -CH-), 5, 45 (1H, dt, J_{AB} = 10.5 Hz, J_{AX} = 7.0 Hz, -CH-); MS (80 eV) m/z (rel. int.) of 11-methyl ester: 238 [M] $^+$ (24), 220 [M $^-$ H $_2$ O] $^+$ (16), 207 [M $^-$ OMe] $^+$ (12), 191 (18), 170 [M $^-$ C $_5$ H $_8$] $^+$ (21), 164 (8), 165 [M $^-$ CCH $_2$)COOMe] $^+$ (7), 151 [M $^-$ (CH $_2$) $_2$ COOMe] $^+$ (75), 133 (13), 121 (9), 109 (40), 97 (67), 83 [C $_5$ H $_7$ O] $^+$ (100).

(1S,2S)-(+)-3-Oxo-2-(2Z-pentenyl)cyclopent-1-yl-butyric acid (12). 0.9 mg; R_t in HPLC (method b): 12.5 min; R_f in TLC systems a (0.27) and b (0.53); $[\alpha]_D^{25} + 48^\circ$; IR $\nu_{\rm max}^{\rm CHCl_3}$ cm $^{-1}$ 3000, 1730, 1652; 1 H NMR (200.13 MHz, CDCl₃, TMS as internal standard): δ 0.95 (3H, t, J = 7.2 Hz, -Me), 1.70–2.75 (16H, m), 5.23 (1H, dt, $J_{\rm AB}$ = 10.5 Hz, $J_{\rm AX}$ = 7.0 Hz, -CH-); 5.44 (1H, dt, $J_{\rm AB}$ = 10.5 Hz, $J_{\rm AX}$ = 7.0 Hz, -CH-); MS (80 eV) m/z (rel. int.) of 12-methyl ester: 252 [M] $^+$ (15), 234 (12), 221 [M $^-$ OMe] $^+$ (5), 196 (10), 184 [M $^-$ C₅H₈] $^+$ (12), 151 [M $^-$ (CH₂)₃COOMe] $^+$ (61), 133 (27), 124 (13), 109 (27), 95 (37), 83 [C₅H₇O] $^+$ (100).

The neutral EtOAc extract was dried (Na₂SO₄), evapd and sepd on CC. Fractions eluted with CHCl₃-EtOAc (95:5) were separated by prep. TLC with system c. Substances between R_f : 0.60-0.73 were recovered which on HPLC (method 2) gave:

Ethyl (+)-7-iso-jasmonic acid [ethyl (1R, 2S)-(+)-3-oxo-2-(2Z-pentenyl)cyclopent-1-yl-acetate] (2). 1.1 mg, R_t in HPLC (method 2): 14.5 min; R_f in TLC system a (0.91) and b (0.68), $[\alpha]_D^{20}$ dextrorotatory; MS (80 eV) m/z (rel. int.) 238 [M]⁺ (23), 220 [M-H₂O]⁺ (6), 209 [M-C₂H₅]⁺ (4), 193 [M-OC₂H₅]⁺ (17), 191 [M-C₂H₅-H₂O]⁺ (12), 170 [M-C₅H₈]⁺ (16), 151 [M-CH₂COOC₂H₅]⁺ (43), 133 (23), 109 (34), 95 (43), 93 (39), 83 [C₅H₇O]⁺ (100), 79 (35); ¹H NMR, IR identical with published data [12] and authentic (-)-6 prepared from (-)-5, EtOH and H₂SO₄.

Catalytic hydrogenation of 9-methyl ester and 13-methyl ester. Adams catalyst (2 mg) in 5 ml EtOH were satd with H₂ and 500 μ g 9-methyl ester or 13-methyl ester added and after 15 min reduction products were recovered giving, in the case of 9-methyl ester, 9, 10-dihydro-1-methyl ester [2, 4] and 9,10-dihydro-5-methyl ester [2, 5] or 10,11-dihydro-13-methyl ester. MS (80 eV) m/z (rel. int.) of the 10,11-dihydro-13-methyl ester: 240 [M] + (5), 207 (7), 170 [M - C₅H₁₀] + (27), 153 [M - (CH₂)₂COOMe] + (43), 109 (9), 96 (32), 83 [C₅H₇O] + (100).

Reduction of 9 with NaBH₄. Compound 9 (200 μ g) was dissolved in a soln of 2 mg NaHCO₃ in 1 ml H₂O and treated with 2 mg NaBH₄ for 30 min at room temp. After acidification to pH 3 with 1 M HCl, extraction with CHCl₃, drying of the CHCl₃ extract with Na₂SO₄ and evapn of the solvent, TLC using solvent system b yielded 40% 3, 40% 4, 10% 7 and 10% 8 [3].

Isomerization of 2, 9, 11 and 12. Compounds 2, 9, 11 and 12 (50 μ g of each) were treated with 100 μ l 1 M KOH. After 1 hr at 60°, 150 μ l 1 M HCl was added and the solns extracted with CHCl₃. TLC with solvent system a yielded 5 (R_f : 0.28) [6], 10 (0.24) [12], 13 (0.30), and 14 (0.32), respectively. The MS of 13 was identical with that of 11, and of 14 with that of 12.

Analytical methods. Methyl esters were prepared by treatment of the carboxylic acids with ethereal CH₂N₂ and analysed by

GC-MS under the following conditions: 80 eV mass spectrometer, glass column (1.80 m × 2 mm) containing 10% EG SS-X on Gas Chrom P (125–150 μ m), column temp. 175°, He 15 ml/min, R_t (min) of the methyl esters: 1 12.6, 3 15.8, 4 13.8, 5 10.3, 7 11.9, 8 11.8; steel column (1.50 m × 2 mm) containing 3% OV-225 on Gas Chrom Q (100–120 mesh), column temp. 170° (6 min)–200° (4°/min); He 17 ml/min, R_t (min) of the methyl esters: 1 4.2, 5 3.5, 9 4.6, 10 4.0, 11 7.0, 12 9.4, 13 6.1, 14 8.3, 9,10-dihydro-1 3.1, 9,10-dihydro-5 3.4; R_t (min): 6 4.0, 2 4.6.

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